

## MW and TM calculation

Oligo Specification Report accompanying all custom oligo synthesis including ALL modifications uses the molecular weights given below.

The molecular weights are not added for modifications and backbone modifications.

ID	Base	BaseAbbreviation	MW	EC	Notes
1	DeoxyAdenosine	A	313.21	15.4	
2	DeoxyCytosine	C	289.19	7.4	
3	DeoxyGuanosine	G	329.21	11.5	
4	Thymidine	T	304.2	8.7	
5	Inosine	I	314.2	7.2	
6	A+G+T+C	N	308.95	10.7	
7	A+G	R	321.21	13.45	
8	C+T	Y	296.69	8.05	
9	A+C	M	301.2	11.4	
10	G+T	K	316.7	10.1	
11	G+C	S	309.2	9.45	
12	A+T	W	308.71	12.05	
13	A+T+C	H	302.2	10.5	
14	G+T+C	B	307.53	9.2	
15	G+A+T	D	315.54	11.86	
16	G+A+C	V	310.53	11.43	
18	phosphate	p	79.98	0	
19	Other	X	0	0	add mw of the
20	deoxy uridine	U	290.17	9.9	

## Formulas

Size= Total number of bases.

%GC= (G+C)/Size

$$mw = (A \times 313.2) + (C \times 289.19) + (G \times 329.21) + (T \times 304.2) + (I \times 314.2) + (N \times 308.95) + (R \times 321.21) + (Y \times 296.69) + (M \times 301.2) + (K \times 316.7) + (S \times 309.2) + (W \times 308.71) + (H \times 302.2) + (B \times 307.53) + (D \times 315.54) + (V \times 310.53) + (P \times 79.98) + (U \times 290.17) - 62$$

$$Tm \text{ For Oligos shorter than 25 bp} = 2(A+T) + 4(C+G)$$

For longer oligos: Reference Bolton, Et and McCarthy, B.J. (1962) PNAS 48: 139-1397

$$Tm = 81.5 - 16.6 + (0.41 \times \%GC) - 600 / \text{size}$$



**At Gene Link the website online ordering software and the oligo report enclosed with the shipment uses the following formula for calculation of melting temperature (Tm) of the oligonucleotide**

$$T_m = 81.5 + 16.6 \times \log_{10}[\text{Na}^+] + 0.41 \times P_{GC} - 600/N$$

Where:  $P_{GC}$  is the total percentage of G and C bases  
N is the sequence length  
 $[\text{Na}^+]$  is 100 mM

Example: 5'- TATATGCCGATCCGCTAGT-3'                    20 mer; GC = 50%; AT = 50%

$$T_m = 81.5 + 16.6 \times \log_{10}[0.100] + 0.41 \times 50 - 600/20$$

$$T_m = 81.5 + 16.6 + 0.41 \times 50 - 600/20$$

$$T_m = 81.5 + 16.6 + 20.5 - 30$$

$$T_m = 64.9 + 20.5 - 30$$

$$T_m = 85.40 - 30$$

$$T_m = 55.4^\circ\text{C}$$

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## Basic

$$T_m = (2 \times N_{AT}) + (4 \times N_{GC})$$

Where:

$N_{AT}$  is the total number of A or T bases

$N_{GC}$  is the total number of C or G bases

Tm for same oligo using  $2(A+T) + 4(C+G)$

$$= 2(5+5) + 4(5+5)$$

$$= 2(10) + 4(10)$$

$$= 20 + 40$$

$$= 60^\circ\text{C}$$



# DNA Extinction Coefficients

Nucleotide (A260 / mol)

A	15.4
T	8.7
G	11.5
C	7.4
I	7.2**
N	10.7*

Neighbors (A260 / mol)

AA	13.7
AT	11.4
AG	12.5
AC	10.6
AI	9.3**
AN	12.2*
TA	11.7
TT	8.4
TG	9.5
TC	8.1
TI	8.1**
TN	9.4*
GA	12.6
GT	10.0
GG	10.8
GC	8.8
GI	8.8**
GN	10.5*
CA	10.6
CT	7.6
CG	9.0
CC	7.3
CI	7.2**
CN	8.6*
IA	9.3**
IT	8.4**
IG	8.8**
IC	7.1**
II	6.8**
IN	8.4*
NA	12.1*
NT	9.4*
NG	9.4*
NC	8.7*
NI	8.7*
NN	9.9*

Handbook of Biochemistry and Molecular Biology (1975) Fasman G.D., ed., 3rd edition, Nucleic Acids - Vol. 1, pp 589, CRC Press, Cleveland, OH.

